TABLE II Drying Times in White Enamel Systems

| Coating formula- tion (see Table I) | Drier concen- tration ^a | Type of drier | Set-to- Touch time (minutes) |
|--|--|------------------------------|---|
| I | .05% Co .05% Co | Naphthenate Neo-decanoate | 34 27 |
| II | .05% Co .05% Co | Naphthenate Neo-decanoate | $\frac{151}{120}$ |
| III | .05% Co .05% Co | Naphthenate Neo-decanoate | 181 169 |
| I | .05% Co + .50% Pb .05% Co + .50% Pb | Naphthenate Neo-decanoate | $\frac{29}{21}$ |
| II | .05% Co + .50% Pb .05% Co + .50% Pb | Naphthenate Neo-decanoate | 133 126 |
| III | .05% Co + .50% Pb .05% Co + .50% Pb | Naphthenate Neo-decanoate | $\begin{array}{c} 118 \\ 107 \end{array}$ |

^a Percentage of metal based on total vehicle solids. ^b Finger-tip touch method (ASTM D-1953).

properties when compared with similar driers based on naphthenic acid. The drying rates shown in Table II were determined with the conventional finger-tip touch method (ASTM D-1953).

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REFERENCES

1. Parker, D. H., "Principles of Surface Coating Technology,"
John Wiley and Sons, New York, 1965, pp. 228-240.
2. Payne, H. F., "Organic Coating Technology," Vol. I, John
Wiley and Sons, New York, 1954, pp. 227-240.
3. Paint Technology Manuals, Part 2, Oil & Colour Chemists'
Association, Reinhold Publishing Corporation, New York, 1961,

pp. 31-57. 4. "Commercial Naphthenate Driers," Nuodex Division of Tenneco

4. Commenced Chemicals Inc.
5. "Neo-decanoic Acid," Enjay Chemical Company.
6. Fefer, M., and A. J. Lauer, JAOCS 44, 331 (1967).

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Color Reactions of Red Palm Oil

Ammonium molybdate-sulfuric acid reagent is reported to give a specific color or turbidity reaction with some fatty oils (1). Experiments at this laboratory revealed two interesting facts (2). Ammonium molybdate does not have any part in the reaction, and an identical response is obtained with concentrated sulfuric acid alone under the same experimental conditions. Red palm oil gives an intense green-blue coloration, which is attributed to the presence of lycopene (3) in the oil. This observation is confirmed by the following.

Bleached palm oil does not give a color in the test because of the destruction of lycopene during the bleaching process. Red palm oil gives the green-blue coloration by direct contact or in nonaqueous media with arsenic trichloride, antimony trichloride, and trichloroacetic acid, which are specific reagents for lycopene, resulting in formation of the unstable carbonium salts (4).

These reagents are rather sensitive both in regard to the size of the sample and the concentration of lycopene. Thus, when a drop of red palm oil is placed on a porcelain tile and a drop (of arsenic trichloride) or crystals of the other two reagents are added and stirred with a glass rod, a blue-green color appears. Peanut oil to which 5% by weight of red palm oil

has been admixed responds to these reagents, especially on a porcelain tile by direct contact.

The causal role of lycopene in the reaction has also been checked by extracting lycopene from fresh and ripe tomatoes with petroleum-ether-acetone and dissolving it (lycopene) in peanut oil and bleached palm oil and performing the test with the use of the four reagents (including concentrated sulfuric acid). A green-blue coloration results in all cases.

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REFERENCES

1. Rajnish Kumar, JAOCS 40, 80 (1963).
2. Ansar Ahmed, S., D. Ramachar, B. A. R. Somayajulu, and S. D. Thirumala Rao, The Oils and Oilseeds J. 19, No. 9 (1966).
3. Deuel, Harry J. Jr., "The Lipids, Their Chemistry and Biochemistry," Vol. 1, Interscience Publishers Inc., New York, 1951, 545.
 Karrer, Paul, "Organic Chemistry," 4th Eng. ed., Elsevier Publishing Company Inc., 1950, p. 708.

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Erratum

"Whiteness and Fluorescence of Fabrics," K. J. Nieuwenhuis, JAOCS 45, 37-42, 1968. On page 39, left-hand column, last line, the "G < 82," should read "G > 32."